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Book Review: **Organic Chemistry** - Paula Y. Bruice

www.organic-chemistry.org/books/reviews/0131963163.shtm

Description. This innovative text is organized in a way that discourages rote memorization, by emphasizing what functional groups do rather than how they are

Books: **Organic Chemistry**

www.organic-chemistry.org/books/navi/organicchemistry.shtm

Advanced **Organic Chemistry** Francis A. Carey, Richard A. Sundberg Paperback, 1199 Pages 5th Edition, 2007 ISBN-13: 978-0-387-44897-8 Springer

Virtual Textbook of **Organic Chemistry: Stereoisomers**

www2.chemistry.msu.edu/faculty/reusch/VirtTxtJml/sterisom.htm

Stereoisomers. As defined in an earlier introductory section, isomers are different compounds that have the same molecular formula. When the group of atoms that

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Chromic acid (H₂CrO₄) as an oxidant in organic chemistry ...

www.masterorganicchemistry.com/.../reagent-friday-chromic-acid-h2cro4

Chromic acid (H₂CrO₄) is a useful oxidant for converting primary alcohols to carboxylic acids, and secondary alcohols to ketones. Here, let's clear up some

What Makes A Good Nucleophile? - Master Organic Chemistry

www.masterorganicchemistry.com/.../06/18/what-makes-a-good-nucleophile

If you read the last post, you'll recall that a **nucleophile** is a species that donates a pair of electrons to form a new covalent bond. Nucleophilicity is measured

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web.chem.ucla.edu/~harding/notes/notes_14C_nmr03.pdf

Diagram 34 *As the applied magnetic field increases, so does the energy difference between - and -spin states <http://www.chemistry.uoguelph.ca/driguana/NMR...>

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